

Vanessa Avila and Edward R. T. Tiekkink*

Department of Chemistry, The University of Texas at San Antonio, One UTSA Circle, San Antonio, Texas 78249-0698, USA

Correspondence e-mail:
edward.tiekkink@utsa.edu

Received 17 November 2006
Accepted 21 November 2006

Key indicators

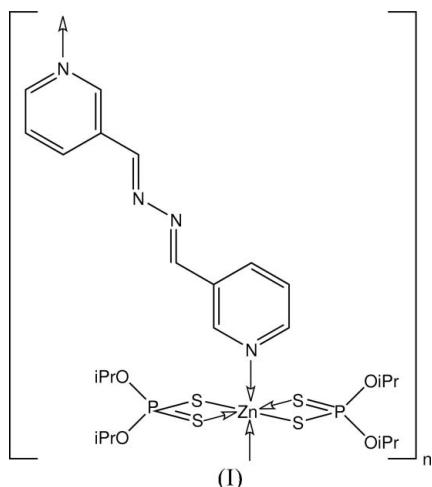
Single-crystal X-ray study
 $T = 150$ K
Mean $\sigma(C-C) = 0.006$ Å
 R factor = 0.044
 wR factor = 0.109
Data-to-parameter ratio = 16.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the polymeric title compound, $[Zn(C_6H_{14}O_2PS_2)_2(C_{12}H_{10}N_4)]_n$, the Zn atom lies on a centre of inversion and the bridging diimine ligand is disposed about another centre of inversion. The Zn atom is in an octahedral geometry defined by a *trans*- N_2S_4 donor set. The polymer topology is that of a step-ladder.

Comment

Structures (Chen *et al.*, 2005; Lai & Tiekkink, 2006) related to the title compound, (I), have been investigated as a part of an ongoing study of the nature of polymer formation in zinc and cadmium dithiophosphates [$\text{S}_2\text{P}(\text{OR})_2$; Lai *et al.*, 2004; Lai & Tiekkink, 2004; Chen *et al.*, 2006]. In (I), the Zn atom exists within a *trans*- N_2S_4 donor set that defines an approximately octahedral geometry. The Zn atom is situated on a centre of inversion and the bis(4-pyridylmethylene)hydrazine is disposed about another inversion centre. The coordination geometry is distorted octahedral defined by a *trans*- N_2S_4 donor set (Fig. 1).

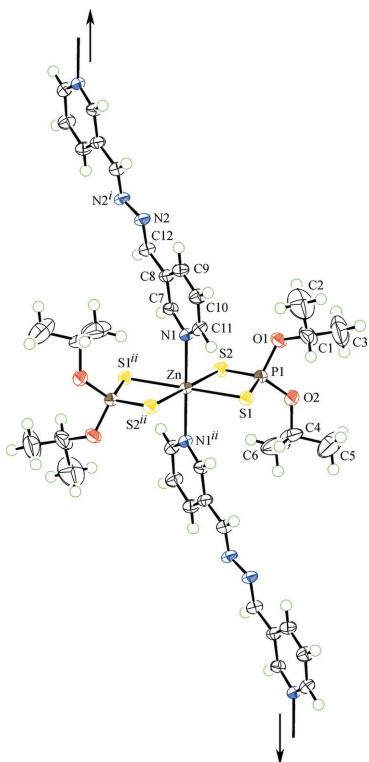


The dithiophosphate ligand coordinates in a symmetric fashion and this coordination mode is reflected in the near equivalence of P—S bond distances and of the Zn—S—P bond angles (Table 1). A view of the polymer in (I) is shown in Fig. 2. The polymer topology, consistent with that found for the isomorphous Cd analogue (Lai & Tiekkink, 2006), is that of a step-ladder.

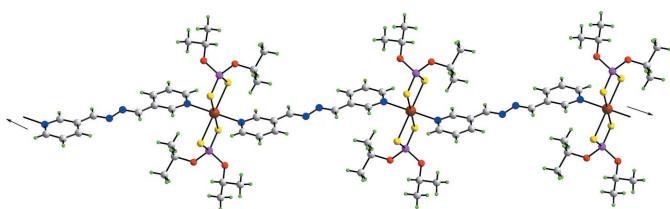
Experimental

© 2006 International Union of Crystallography
All rights reserved

Compound (I) was prepared by refluxing the parent $Zn[S_2P(OiPr)_2]_2$ compound with bis(3-pyridylmethylene)hydrazine (Aldrich)

**Figure 1**

Part of the polymeric structure of (I), showing the crystallographic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry codes: (i) $1-x, -y, 2-z$; (ii) $-x, -y, 1-z$]

**Figure 2**

Polymer topology in (I). Colour code: Zn brown, S yellow, P pink, O red, N blue, C grey and H green.

according to a literature procedure (Lai & Tiekkink, 2004). Yellow crystals were isolated by the slow evaporation of an 2-propanol solution of the compound (m.p. 381–384 K). IR (KBr disk): $\nu(\text{C}=\text{N})$ 1464 (*w*), $\nu(\text{C}-\text{O})$ 1101 (*w*), $\nu(\text{P}-\text{O})$ 960 (*s*), $\nu(\text{P}-\text{S})_{\text{asym}}$ 646 (*s*) cm^{-1} .

Crystal data

[Zn(C₆H₁₄O₂PS₂)₂(C₁₂H₁₀N₄)]

$M_r = 702.13$

Monoclinic, $P2_1/n$

$a = 11.809$ (3) Å

$b = 11.243$ (3) Å

$c = 12.933$ (3) Å

$\beta = 108.712$ (5)°

$V = 1626.3$ (7) Å³

$Z = 2$

$D_x = 1.434 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

$\mu = 1.15 \text{ mm}^{-1}$

$T = 150$ (2) K

Plate, yellow

0.30 × 0.25 × 0.08 mm

Data collection

AFC12K/SATURN724

diffractometer

ω scans

Absorption correction: multi-scan

(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.722$, $T_{\max} = 1$

(expected range = 0.659–0.912)

34154 measured reflections

2864 independent reflections

2793 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$

$\theta_{\max} = 25.0^\circ$

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.044$

$wR(F^2) = 0.109$

$S = 1.08$

2864 reflections

178 parameters

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0527P)^2$

+ 4.0311P]

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\max} = 1.46 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.57 \text{ e } \text{\AA}^{-3}$

Table 1
Selected geometric parameters (Å, °).

Zn–S1	2.5420 (9)	S2–P1	1.9792 (11)
Zn–S2	2.5540 (9)	N2–N2 ⁱ	1.412 (5)
Zn–N1	2.228 (2)	N2–C12	1.268 (4)
S1–P1	2.0013 (11)		
S1–Zn–S2	81.23 (3)	S2–Zn–N1	92.20 (7)
S1–Zn–N1	89.65 (7)	S2–Zn–N1 ⁱⁱ	87.80 (7)
S1–Zn–S2 ⁱⁱ	98.77 (3)	Zn–S1–P1	82.50 (3)
S1 ⁱⁱ –Zn–N1	90.35 (7)	Zn–S2–P1	82.61 (4)

Symmetry codes: (i) $-x + 1, -y, -z + 2$; (ii) $-x, -y, -z + 1$.

H atoms were included in the riding-model approximation, with $\text{C}-\text{H} = 0.95\text{--}1.00 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{methyl C})$ or $1.2U_{\text{eq}}(\text{C})$. The highest peak is located 1.36 Å from atom C4.

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *PATTY* in *DIRDIF92* (Beurskens *et al.*, 1992); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *SHELXL97*.

Cheminova is thanked for the gift of the dithiophosphate ligand used in this study.

References

- Beurskens, P. T., Admiraal, G., Beurskens, G., Bosman, W. P., Garcia-Granda, S., Gould, R. O., Smits, J. M. M. & Smykalla, C. (1992). *The DIRDIF Program System*. Technical Report. Crystallography Laboratory, University of Nijmegen, The Netherlands.
- Brandenburg, K. (2006). *DIAMOND*. Release 3.1. Crystal Impact GbR, Bonn, Germany.
- Chen, D., Lai, C. S. & Tiekkink, E. R. T. (2005). *Acta Cryst. E* **61**, m2052–m2054.
- Chen, D., Lai, C. S. & Tiekkink, E. R. T. (2006). *CrystEngComm*, **8**, 51–58.
- Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Lai, C. S., Liu, S. & Tiekkink, E. R. T. (2004). *CrystEngComm*, **6**, 221–226.
- Lai, C. S. & Tiekkink, E. R. T. (2004). *CrystEngComm*, **6**, 593–605.
- Lai, C. S. & Tiekkink, E. R. T. (2006). *Z. Kristallogr.* **221**, 288–293.
- Rigaku (2005). *CrystalClear*. Rigaku/MSC Inc., The Woodlands, Texas, USA.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.